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Key indicators

Single-crystal X-ray study T = 292 KMean $\sigma(\text{C}-\text{C}) = 0.004 \text{ Å}$ R factor = 0.058 wR factor = 0.149 Data-to-parameter ratio = 17.5

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

3-Benzyl-2-butylamino-5,6,7,8-tetrahydrobenzothieno[2,3-*d*]pyrimidin-4(3*H*)-one

In the title molecule, $C_{21}H_{25}N_3OS$, the fused thiophene and pyrimidine rings are essentially coplanar. In the crystal structure, $N-H\cdots O$ and weak $C-H\cdots O$ hydrogen bonds link the molecules, forming one-dimensional chains along [010].

Comment

Pyrimidine derivatives are currently attracting attention because of the important role played by such systems in many natural products and antibiotics (Ding *et al.*, 2004). We have been interested in the preparation of derivatives of heterocycles *via* the aza-Wittig reaction. We report here the synthesis and structure of the title compound, (I).

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In the title molecule (Fig. 1), all ring atoms in the thieno [2,3-d] pyrimidine system are essentially coplanar, indicating that the thieno [2,3-d] pyrimidine system is conjugated. There exists



Figure 1

View of the molecular structure of (I), showing displacement ellipsoids drawn at the 50% probability level. H atoms are represented as spheres of arbitrary radii.

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Figure 2



a crossed angle between the benzene ring and the above mentioned plane, and no disorder of the molecule. In the crystal structure, N-H···O and weak C-H···O hydrogen bonds link the molecules, forming one-dimensional chains along [010] (Fig. 2 and Table 2).

Experimental

To a solution of iminophosphorane (1.45 g, 3 mmol) in anhydrous dichloromethane (15 ml), butyl isocyanate (3 mmol) was added under dry N₂ at room temperature. After the reaction mixture was left unstirred for 48 h at room temperature, the solvent was removed under reduced pressure and diethyl ether/petroleum ether (1:2 v/v, 20 ml) was added to precipitate triphenylphosphine oxide. After filtration the solvent was removed to give the carbodiimide, which was used directly without further purification. To a solution of the carbodiimide (15 ml), benzylamine (3 mmol) was added. After the mixture was stirred for 6 h, the solvent was removed and anhydrous ethanol (10 ml) containing several drops of EtONa in EtOH was added. The mixture was stirred for 12 h at room temperature. The solution was concentrated and the residue was recrystallized from ethanol to give the title compound (I), in a yield of 47% (m.p. 379 K). ¹H NMR (CDCl₃, 400 MHz): δ 7.36–7.24 (*m*, 5H, Ar–H), 5.25 (*s*, 2H, Ph-CH2), 4.35 (br, 1H, NH), 3.32-3.30 (m, 2H, NCH2), 2.99-2.69 (m, 4H, 2CH2), 1.86-1.84 (m, 4H, 2CH2), 1.39-1.35 (m, 2H, CH2), 1.15–1.05 (*m*, 2H, CH2), 0.82–0.78 (*t*, *J* = 7.4 Hz, 3H, CH3); MS (*m*/*z*, %), 367(*M*+, 7), 277 (2), 104 (3), 91 (100), 77 (3), 65 (17), 57 (7). Elemental analysis calculated for C21H25N3OS: C 68.63, H 6.86, N 11.43%; found: C 68.71, H 7.93, N 11.34%. Crystals suitable for single-crystal X-ray diffraction were obtained from hexane and dichloromethane at room temperature.

Crystal data

C21H25N3OS	$D_{\rm r} = 1.256 {\rm Mg} {\rm m}^{-3}$
$M_r = 367.50$	Mo $K\alpha$ radiation
Monoclinic, $P2_1/c$	Cell parameters from 1789
a = 11.7759 (11) Å	reflections
b = 9.2747 (9) Å	$\theta = 2.3-22.4^{\circ}$
c = 18.6871 (18) Å	$\mu = 0.18 \text{ mm}^{-1}$
$\beta = 107.760 \ (2)^{\circ}$	T = 292 (2) K
V = 1943.7 (3) Å ³	Block, colorless
Z = 4	$0.30 \times 0.20 \times 0.10 \text{ mm}$

Data collection

Bruker SMART CCD area-detector diffractometer	4192 independent reflections 2809 reflections with $I > 2\sigma(I)$
φ and ω scans	$R_{\rm int} = 0.039$
Absorption correction: multi-scan	$\theta_{\rm max} = 27.0^{\circ}$
(SADABS; Sheldrick, 1996)	$h = -13 \rightarrow 15$
$T_{\min} = 0.948, \ T_{\max} = 0.982$	$k = -11 \rightarrow 11$
11061 measured reflections	$l = -23 \rightarrow 21$
Refinement	
Refinement on F^2	$w = 1/[\sigma^2(F_0^2) + (0.0688P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.058$	+ 0.0324P]
$wR(F^2) = 0.149$	where $P = (F_0^2 + 2F_c^2)/3$
S = 1.04	$(\Delta/\sigma)_{\rm max} = 0.002$

4192 reflections 240 parameters H atoms treated by a mixture of independent and constrained refinement

809 reflections w	ith $I > 2\sigma(I)$
$R_{int} = 0.039$	
$max = 27.0^{\circ}$	
$s = -13 \rightarrow 15$	
$z = -11 \rightarrow 11$	
$= -23 \rightarrow 21$	

$r^{2}(F_{0}^{2}) + (0.0688P)^{2}$ 0324P] $e P = (F_0^2 + 2F_c^2)/3$ $\Delta/\sigma)_{\rm max} = 0.002$ $\Delta \rho_{\rm max} = 0.21 \text{ e } \text{\AA}^{-3}$ $\Delta \rho_{\rm min} = -0.19 \text{ e } \text{\AA}^{-3}$

Selected geometric parameters (Å, °).

Table 1

C5-C6	1.351 (3)	C8-S1	1.734 (2)
C5-S1	1.742 (2)	C9-O1	1.233 (2)
C6-C7	1.436 (3)	C9-N1	1.406 (3)
C7-C8	1.390 (3)	C17-N2	1.314 (3)
C7-C9	1.422 (3)	C17-N1	1.392 (3)
C8-N2	1.348 (3)		
C8-S1-C5	91.56 (10)		

Table 2			
Hydrogen-bond	geometry (.	Å, °΄).

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
$C10-H10B\cdotsO1^{i}$ $N3-H3\cdotsO1^{i}$	0.97	2.40	3.370 (3)	174
	0.84 (1)	2.12 (1)	2.928 (2)	162 (2)

Symmetry code: (i) -x + 1, $y + \frac{1}{2}$, $-z + \frac{3}{2}$.

The H atom bonded to N3 was refined isotropically. All other H atoms were placed in calculated positions and treated as riding atoms, with C-H = 0.93–0.97 Å and $U_{iso}(H) = 1.2U_{eq}(C)$ or $1.5U_{eq}(methyl)$ C).

Data collection: SMART (Bruker, 2001); cell refinement: SAINT (Bruker, 2001); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: PLATON (Spek, 2003); software used to prepare material for publication: SHELXTL (Sheldrick, 2001).

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